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(74) Agent: PERSLEY, Sidney; 222 Bridge Plaza South, Fort Lee, NJ 07024 (US).

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(54) Title: PREPARATION OF HEAT STABLE PIGMENT VIOLET 23

(57) Abstract: A process for improving the heat stability of Pigment Violet 23 by conditioning crude Pigment Violet 23 in the presence of about 1 to 10 wt.% of Pigment Violet 23 by-products (1, 2 and 3).

PREPARATION OF HEAT STABLE PIGMENT VIOLET 23

5 FIELD OF THE INVENTION

The invention relates to an improved process for preparing a heat stable pigment violet 23.

BACKGROUND OF THE INVENTION

Pigment violet 23 has been known for decades and in the 10 course of this time has acquired a very great importance because of its outstanding properties as a colorant. It is therefore manufactured on an industrial scale in large quantities. As long as the colorant has been known, the principal features of its manufacture have remained the same. It is typically synthesized 15 by reacting chloranil (tetrachloro-p-benzoquinone) with 3-amino-9-ethylcarbazole to give 2,5-di-(9-ethylcarbazol -3-ylamino)-3,6-dichloro-1, 4-benzoquinone. This is followed by cyclisation to give the pigment violet 23. (See Venkataraman, The Chemistry of Synthetic Dyes, Volume II (1952), pages 786 and 20 787). The following equations serves to illustrate the synthetic reaction scheme:

5
$$Cl$$
 Cl $+2$ NH_2 Acid acceptor $-2HCl$

10 C_2H_5 Cl H Cl C_2H_5 C_2H_5

Pigment Violet 23

While many variations in the solvents used and conditions of this reaction have been disclosed, invariably, the reaction also leads to the formation of several by-products that the prior art deemed contaminants. These by-products are normally washed away from the crude pigment violet 23 before any conditioning (i.e. heat stabilizing) of such crude occurs.

SUMMARY OF THE INVENTION

The present invention provides an improvement in the process for the conditioning of pigment violet 23 having the formula:

5

the improvement comprises carrying out said conditioning in the presence of about 1 to about 10 wt.%, preferably about 5 to 8 wt.%, relative to the weight of the crude pigment, of at least one compound selected from the group consisting of:

15

$$\begin{array}{c|c} Cl & Et \\ \hline \\ SC_6H_4R & \end{array}$$

20

(2); and

25

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30 wherein each R is independently selected from either H or CH3.

violet 23 product is conditioned in the presence of about 1-10 wt.% of at least one of the following compounds:

wherein each R is independently selected from either H or CH3.

The conditioning of crude pigment violet 23 may be carried

25 out by various processes such as by grinding the crude pigment
in the presence of a grinding agent and an organic solvent.

Preferably, the grinding agent is sodium chloride and the
organic solvent is diethylene glycol.

Compounds 1, 2 and 3 are formed in final step in commercial preparation of pigment violet 23 crude which involves ring closure of the intermediate di-anil (4)

using benzenesulfonyl chloride or p-toluenesulfonyl chloride (or other arylsulfonyl chlorides).

It is well known that cyclization of the di-anil

intermediate (4) during the synthesis of crude violet 23 results in the concurrent formation of by-product compounds (1), (2) and (3). More specifically, in compounds (1), (2) and (3), R is H when benzenesulfonyl chloride is employed as the cyclizing agent and R is CH₃ when p-toluenesulfonyl chloride is employed as the cyclizing agent. Alternatively, a mixture of (1), (2) and (3) can be produced wherein R is a mixture of H and CH₃.

20 This is the case when a mixture of benzenesulfonyl chloride and p-toluenesulfonyl chloride are employed as cyclization agents. When aryl sulfonyl chlorides including other isomers of toluenesulfonyl chlorides, naphthalenesulfonyl chlorides, anthracenesulfonyl chlorides or other aryl systems are employed as the cyclization agents, the corresponding by-products are

The amount of these by-products [i.e. compounds; (1), (2), and (3)] retained in the isolated crude product may be controlled by the washout procedure applied during the isolation/filtration process. Thus, rigorous washing with the hot reaction solvent will remove virtually all of the by-product

also formed and result in the same heat stabilizing effect.

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less rigorous washing conditions with preferably 1-10 wt.%
retained. Alternatively, the appropriate level of (1), (2) or
(3) by-product compounds or mixtures thereof (hereinafter
referred to as "by-product mixture"), may be added to a purified

10 crude sample prior to its conditioning. The presence of at
least one of compounds (1), (2) or (3) in an amount of about 110 wt.% causes significant inhibition of recrystallization
during the conditioning process. The result is that pigments
possessing these levels of by-products (or impurities) also

exhibit correspondingly higher surface areas.

For example, when a crude product containing <1 wt.% of at least one of compounds (1), (2) or (3) is conditioned, the resulting surface area is typically in the range of 85 to 95

20 m²/g. By contrast, when a crude product containing from about 1 to 10 wt.%, preferably from about 5 to 8 wt.% of at least one of compounds (1), (2) or (3) is conditioned, the resulting surface area is typically in the range of 115 to 125 m²/g. The superior heat stability of the latter product in plastic, compared to the former is also due to the presence of one or more of these compounds. Thus, crystalline growth (or recrystallization) of the pigment is significantly inhibited, during the high temperature processing necessary for the coloration of plastics.

30 The examples that follow demonstrate that the heat stability exhibited by plastic colored with the pigment prepared from a crude pigment violet 23 containing from 5 to 10 wt.% of

the equivalent plastic colored with the pigment prepared from crude pigment containing <1 wt.% of the same compound(s).

Example 1

10 Preparation of the Carbazole Violet crude was carried out as described in U.S. Patent No. 4,345,074 with the exception that upon completion of the synthetic process, the reaction mixture was cooled to 90 C and filtered. The dichlorobenzene-saturated filter cake was slurried in methanol, re-filtered, washed with methanol then with water. After drying, the crude Carbazole Violet was conditioned (i.e. pigmented) by various well known procedures as described in U.S. Patent No. 4,345,074. Such procedure resulted in a Pigment Violet 23 containing 5 to 10 wt.% of the mixture of (1), (2) and (3) which exhibited increased heat stability when dispersed in polyethylene or polypropylene, relative to the product generated by Example 2.

Example 2

(Comparative)

25 Preparation of the Carbazole Violet crude was carried out as described in Example 1. The reaction mixture was cooled to 100 C, filtered hot, rinsed with o-dichloro-benzene, then warmed to 100 C until the filtrates became colorless. The filter cake was then slurried in o-dichlorobenzene, warmed to 170 C for 1 hour, then cooled to 100 C and filtered. The filtrate was then rinsed with additional hot dichlorobenzene until the filtrate

dichlorobenzene, and finally with 50 C water. The wet filtercake was dried in an oven. The crude was pigmented by various procedures as described in U.S. Patent No. 4,345,074. These procedure resulted in a pigment containing <1 wt.% of a mixture of by-product compounds (1), (2) and (3) and exhibited decreased heat stability when dispersed in polyethylene or polypropylene relative to the product generated by Example 1.

In a high-speed, high-shear/temperature polypropylene dispersion system, the product (Example 2) from the pigment 15 containing <1 wt.% of at least one of compounds (1), (2) or (3) exhibited a loss in color strength of approximately 10% when compared, under identical conditions, to the equivalent product (Example 1) prepared from the pigment containing 5 to 10 wt.% of at least one of compounds (1), (2) or (3), during a processing 20 period of 30 seconds to 180 seconds. Additionally, the coloristic properties of the product in Example 2 were more adversely affected. Specifically, Da and Db of the product in Example 2 changed by -2.5 units and +1.4 units, respectively, relative to the corresponding color coordinates of the product in Example 1. Therefore, during the process of plastic compounding, a trend is observed. The products obtained from the less stabilized pigments were greener, yellower and weaker and therefore less desirable commercially than the products obtained from the more stabilized pigments.

Example 3

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(3) were added to nine parts of a purified carbazole violet crude. To this was added 100 parts of sodium chloride grinding agent and 18 parts of diethylene glycol. The material was mixed in a kneader for 5 hours, then slurried in 1000 parts of water at 85 to 90 C for one hour. The slurry was filtered, rinsed with water, then oven dried. The resulting pigment had increased heat stability comparable to that of the product from Example 1.

The invention has been described in terms of preferred embodiments thereof, but is more broadly applicable as will be understood by those skilled in the art. The scope of the invention is only limited by the following claims.

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1. A process for improving the heat stability of Pigment Violet 23 having the formula:

comprising conditioning said pigment in the presence of about 1 to about 10 wt.%, relative to the weight of the crude pigment, of at least one compound selected from the group consisting of:

20
$$\begin{array}{c}
CI \\
N \\
CI \\
N \\
SC_6H_4R
\end{array}$$

$$\begin{array}{c}
Et \\
SC_6H_4R
\end{array}$$

30 wherein each R is independently selected from either H or CH_3 .

10

10

2. The process of claim 1 wherein the conditioning is carried out by grinding the crude pigment in the presence of a grinding agent and an organic solvent.

- 3. The process of claim 2 wherein the grinding agent is sodium chloride.
 - 4. The process of claim 2 wherein the organic solvent is diethylene glycol.

- 5. The process of claim 1 wherein carrying out said conditioning is carried out in the presence of about 5 to 8 wt.% of at least one of said compounds (1), (2) or (3).
- 20 6. A process for increasing surface area of Pigment Violet 23 comprising conditioning said pigment according to the process of claim 1.
- 7. A heat stable Pigment Violet 23 prepared according to the process of claim 1.

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A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C09B67/22 C09B67/04

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B. FIELDS SEARCHED

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Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, CHEM ABS Data, WPI Data

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Further documents are listed in the continuation of box C.	X Patent family members are listed in annex.
Special categories of cited documents: A* document defining the general state of the art which is not considered to be of particular relevance E* earlier document but published on or after the international filing date L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) O* document referring to an oral disclosure, use, exhibition or other means P* document published prior to the international filing date but later than the priority date claimed	 *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. *&* document member of the same patent family
Date of the actual completion of the international search	Date of mailing of the international search report
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Name and mailing address of the ISA	Authorized officer
European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Dauksch, H

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